

**Biology 5357**  
**Chemistry & Physics of Biomolecules**  
**Examination #1**

Proteins Module

September 27, 2019

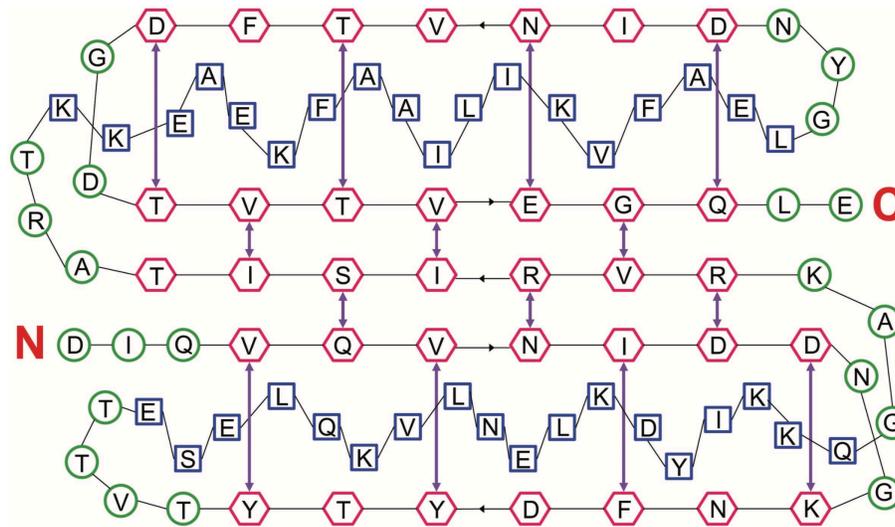
Name: \_\_\_\_\_

**Question 1 (10 points)**

- (A) The widely-used non-nutritive sweetener aspartame (a.k.a. “NutraSweet”), is the C-terminal methyl ester of L-Asp-L-Phe. Draw the chemical structure of aspartame in neutral aqueous solution, paying attention to protonation state and chirality. (Remember an ester is a carboxylic acid derivative, and a methyl ester has the formula R-COOCH<sub>3</sub>.)
- (B) In the protease subtilisin, removal of a negative charge via mutagenesis changes the pK<sub>a</sub> of a His residue located some 15 Å away by about 0.4 units. Use this information to estimate an “effective dielectric constant” which measures the shielding between these charges located at the protein surface.

**Question 2 (15 points)**

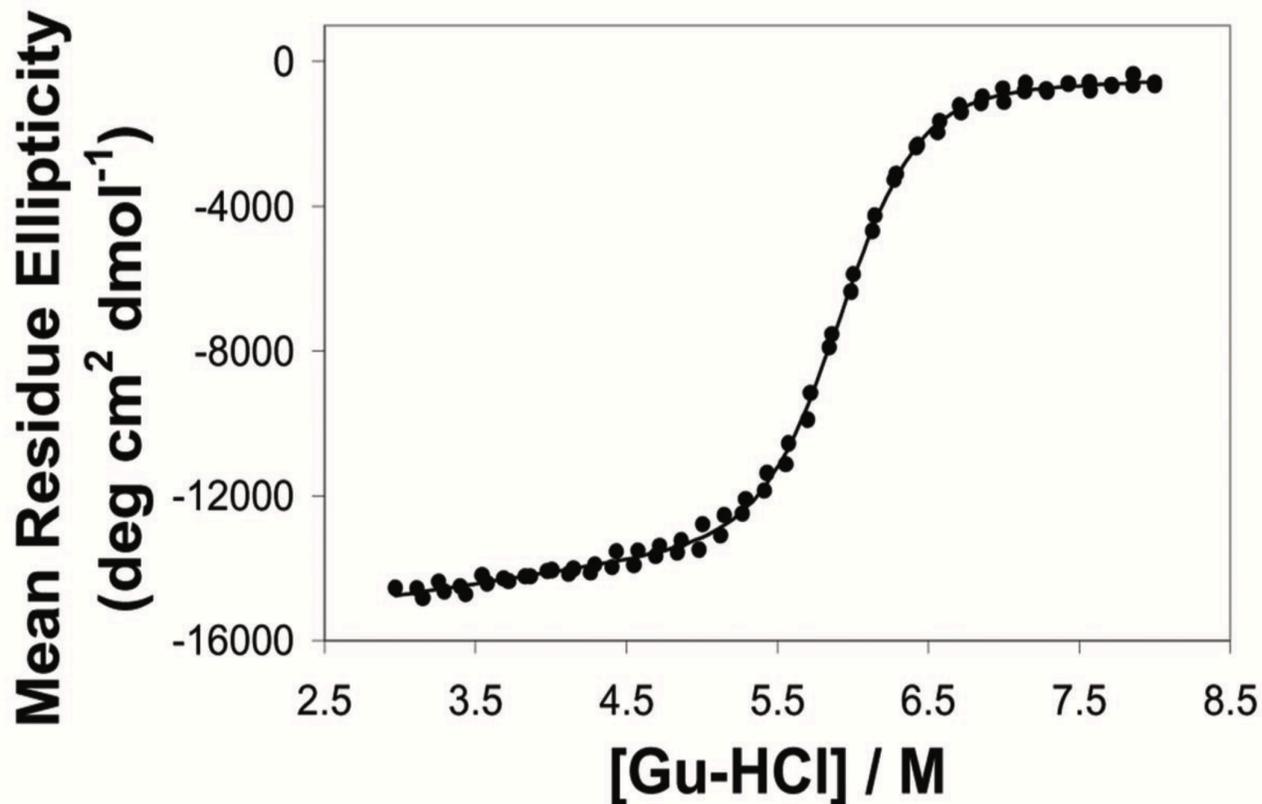
Shown below is a schematic diagram of the amino acid sequence of the TOP7 protein designed by the David Baker lab at the University of Washington using their ROSETTA software.



- (A) Is the TOP7  $\beta$ -sheet structure parallel, antiparallel, or a mixture of the two?
- (B) There are two  $\alpha$ -helices, one closer to the N-terminus and another closer to the C-terminus. For each helix, indicate whether it lies above or below the plane of the page, assuming the  $\beta$ -sheet lies roughly in the plane of the page.
- (C) What is the definition of a  $\beta$ -turn? Circle the residues comprising one of the  $\beta$ -turns in the TOP7 structure.

### Question 3 (15 points)

Provided below is the equilibrium folding-unfolding curve for the TOP7 protein from the previous question as measured at room temperature ( $25^{\circ}\text{C} = 298\text{K}$ ). The plot shows circular dichroism (CD) signal at 222 nm as a function of guanidinium chloride (Gu-HCl) concentration. Use this data to estimate the  $\Delta G$  between the folded and unfolded states of TOP7. (Note  $R = 1.987 \text{ cal/mol/K}$ )

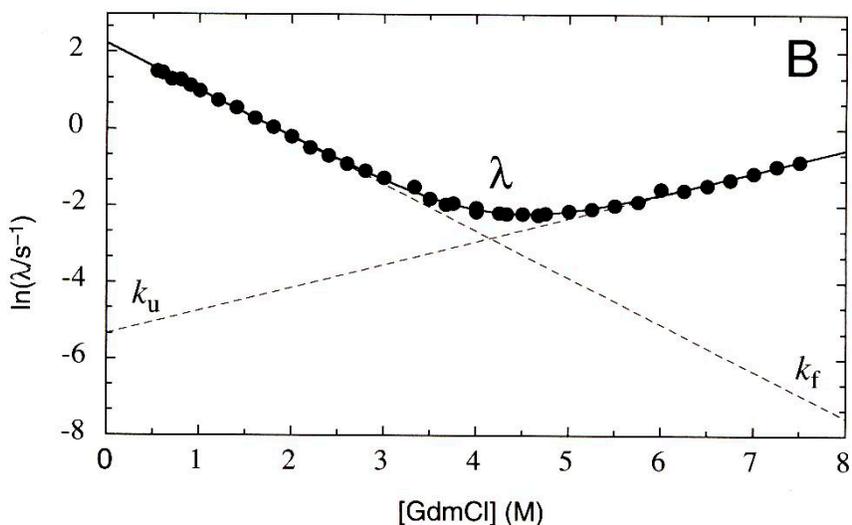


**Question 4 (15 points)** A synthetic 16-residue peptide capped with *N*-acetyl at the N-terminus, Ac-KKYTVSINGKKITVSI, has been shown to reversibly fold into a  $\beta$ -hairpin structure. NMR experiments monitored the chemical shift of the  $H_{\alpha}$  protons as a function of temperature in both pure water and 50% methanol. Analysis of the data gives the thermodynamic values shown below for the folding process. All values are corrected to 25°C (298K), and are given in kJ/mol for  $\Delta H$ , and J/mol/K for  $\Delta S$  and  $\Delta C_p$  (1 cal = 4.184 J, R = 1.987 cal/mol/K).

Solvent	$\Delta H^{\circ}$	$\Delta S^{\circ}$	$\Delta C_p^{\circ}$
Pure H <sub>2</sub> O	+7.2	+23	-1400
50% CH <sub>3</sub> OH	-38.5	-114	-11

- (A) Briefly explain the data analysis procedures used to derive these thermodynamic values from the experimental data.
- (B) What percentage of the time is the peptide in the  $\beta$ -hairpin conformation in each of these two solvents?
- (C) All three of the thermodynamic parameters undergo a change with solvent composition. Provide a physical reason for the observed shift in each value.

**Question 5 (10 points)** Tendamistat is a 74-residue  $\beta$ -sheet protein that specifically inhibits mammalian  $\alpha$ -amylases via formation of a 1:1 complex. Below is data from kinetic folding-unfolding experiments performed at room temperature and pH = 2.



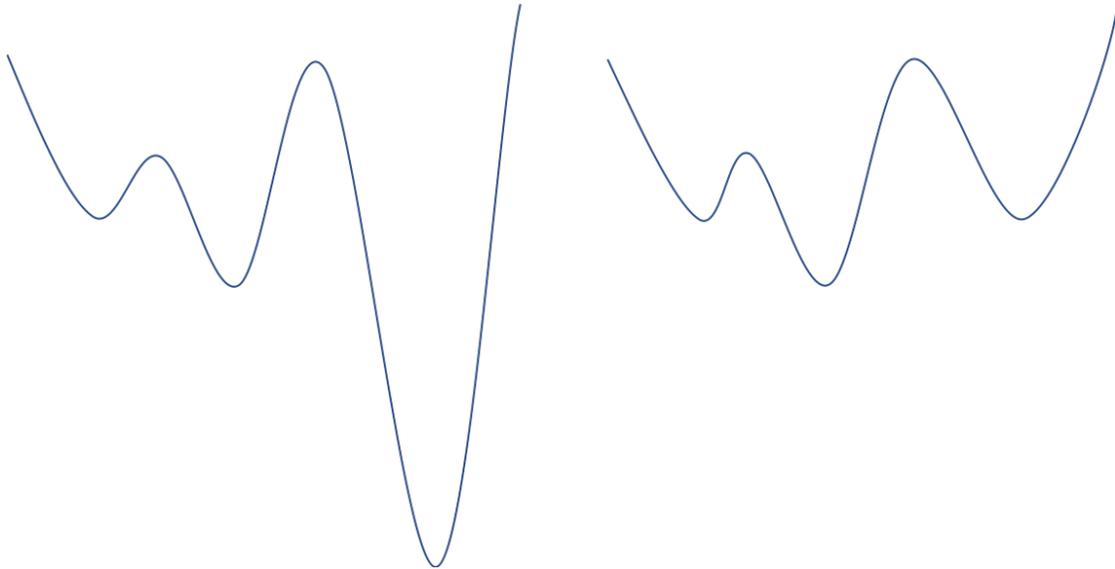
- (A) From the kinetic data, estimate the stability of the folded form of tendamistat relative to the unfolded form in the absence of denaturant.
- (B) Kinetic data is often treated via conventional transition state theory which connects the rate constant ( $k$ ) to the free energy barrier ( $\Delta G^\ddagger$ ) via the relation

$$k = \left( \frac{k_B T}{h} \right) \cdot e^{-\frac{\Delta G^\ddagger}{RT}}$$

where  $k_B$  is Boltzmann's constant ( $1.38 \times 10^{-23} \text{ m}^2 \text{ kg s}^{-2} \text{ K}^{-1}$ ),  $h$  is Planck's constant ( $6.626 \times 10^{-34} \text{ m}^2 \text{ kg s}^{-1}$ ), and  $R$  is the gas constant ( $1.9872 \text{ cal/mol/deg}$ ). Draw a free energy vs. "reaction coordinate" diagram for the folding process. Find the energy of the folding transition state.

**Question 6 (10 points)**

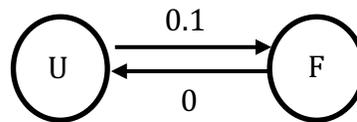
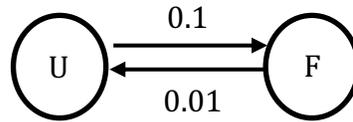
- (A) Which landscape is more complex according to the Shannon entropy metric?  
Circle your answer.



- (B) Explain what it is that makes the landscape you chose in part (A) more complex?

**Question 7 (10 points)**

- (A) The following Markov state models (MSMs) have an unfolded state (U), a folded state (F), and transition probabilities between them. Which one violates detailed balance? (circle one)



- (B) What's "wrong" with the model that violates detailed balance?

**Question 8 (5 points)**

You are studying the folding of a protein that appears to have two-state behavior. The equilibrium probability of the folded state is 0.9 and the equilibrium probability of the unfolded state is 0.1. You've also determined that the probability of unfolding in a 100 ms time interval is 0.005. What is the probability of folding in a 100 ms time interval?

**Question 9 (10 points)**

Which coarse-graining is a better representation of the black energy landscape according to the relative entropy metric? (your choices are blue dotted or red dashed)  
Why is your choice a better representation?

